# COSHA MAP

#### SAMRAT ASHOK TECHNOLOGICAL INSTITUTE

#### (Engineering College), VIDISHA M.P.

(An Autonomous Institute Affiliated to RGPV Bhopal)

# Department of Applied Science SYLLABUS FOR CE, ME and AE Programs

Subject Category	BSC	Subje Cod	' CHR	102	Subject Name:	Chemistry						
	Maximum Marks Allotted								Contact Hours Total			
	11	neory		Pra	actical	Total Marks				Credits		
End Sem	Mid-S	em C	Quiz/Assignment	End Sem	Lab-Work	Total Warks	L	Т	Р	Credits		
60	20		20	30	20	150	3	-	2	4		

#### Prerequisites:

Students who have completed 12th with Science stream or Chemistry of 12th standard or equivalent

#### Course Objective:

The main aim of Engineering Chemistry is to make Students familiar with basic concepts of Chemistry, the students face in industry and engineering field. With this background the Students will be able to explain Scientifically the various chemistry related problems in industry/engineering field.

#### Course Outcomes:

Student after successful completion of course shall possess skills to think critically and analyse chemistry problems in engineering field. Students are expected to solve the chemistry problems with an engineering purview. Laboratory work is intended for students to learn conducting experiments and analyse experimental data.

Sno.	CO's Description								
CO1	Differentiate hard & soft water, solve the related numerical on water treatment and have								
	knowledge regarding its Significance in industry and daily life.								
CO2	pply their knowledge regarding various types of fuels including petroleum fuels, Fuels Cells,								
	Electrical Vehicle Batteries								
CO3	Acquire basic knowledge of various types of polymers, with mechanism and applications.								
CO4	To know basic concept of lubrication and its properties. To have knowledge about cement and								
	refractories to appreciate its applications in various industries.								
CO5	Analyze the need of instruments. Identify and estimate about the unknown/new compounds with								
	the help of spectroscopy/ chromatography.								

UN ITs	Descriptions	Hrs.	CO's	Re mar ks
I	WATER TECHNOLOGY: Sources, Availability, impurities in Water, Types of hardness, Units of hardness. Concentration expression: Normality, Molarity, Molality. Water analysis techniques  – Hardness determination by EDTA method, Alkalinity determination. Defects in boiler due to Hard water. External Treatment (Lime-soda, Zeolite & Ion exchange resin method) & Internal Treatment of Boiler feed water. Numerical Problems.	8	1	
11	FUELS & ENERGY STORAGE SYSTEMS: Characteristics of fuels. Classification of fuels, Calorific Value, HCV, NCV. Proximate and ultimate analysis of coal. Petroleum & its refining. Knocking, Octane Number & Cetane Number, Numerical problems. Electrochemistry: Introduction, EMF of cell, Single electrode potential. Classification of batteries (primary, secondary and reserved batteries), Introduction to Fuel Cell, Electrical Vehicle Batteries their components and materials used.	8	2	
III	POLYMERS AND NANOMATERIALS: Polymers: Nomenclature & classification of polymers. Thermoplastics and	8	3	

	The second secon								
	Thermosetting polymers. Preparation, properties and applications of PE, PVC, PS,								
	Teflon, Nylon 6:6, PU, SBR, NBR, Bakelite, Silicone resin. Rubber and its types.								
	Vulcanization of Rubber, Applications of rubber.								
	Photoactive polymers, Photovoltaic materials: solar cells and dye sensitized solar								
	cells- principle and applications.								
	Nanomaterials: Introduction, Synthesis and applications of nano materials.								
	Introduction to smart materials and its application.								
	LUBRICANTS AND CEMENTING MATERIAL:								
	Introduction, Classification &functions, Mechanism of lubrication, Lubricating oils,								
	grease, semisolid lubricant and solid lubricants. Properties of lubricating oils with	_	.						
IV	significance: Viscosity Index, Flash point, Fire point, Aniline point, Cloud & pour	8	4						
	point, Steam EmulsionNumber (S.E.N), Numerical problems.								
	Composition of Cement, Manufacture of Portland cement. Chemistry of Setting and								
	hardening of cement.								
	INSTRUMENTAL METHODS OF ANALYSIS:								
	Importance of Instrumental techniques. Classification of Instrumental techniques.								
V	Introduction to Electroanalytical and Spectroscopic Methods.Principle,	8	5						
"	Instrumentation, Working and applications of following techniques: Colorimetry, IR	Ŭ							
	Spectroscopy, Conductometry, pH metry, Chromatography and Gas								
	Chromatography.								
Gues	st Lectures (if any)								
Tota	l Hours	40							
_									

#### Suggestive list of experiments:

#### LABORATORY EXPERIMENTS: (Any 10 experiments to be performed)

- 1. To determine strength of unknown Ferrous Ammonium Sulphate FeSO<sub>4</sub>.(NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>.6H<sub>2</sub>O (Mohr's Salt) solution by titrating it against intermediate Potassium Dichromate (K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>) solution using Di Phenyl Amine(DPA) as internal indicator.[Redox Titration]
- 2. To determine Temporary, Permanent and Total Hardness in given sample of water by E.D.T.A. method.[Complexometric Titration]
- To determine strength of Sodium Carbonate and Sodium Bicarbonate in given alkaline solution by titrating with standard HCl using phenolphthalein and Methyl Orange indicators.
  - To determine alkalinity in given water sample using Phenolphthalein and Methyl Orange indicators.[Acid Base Titration]
- 4. To determine strength of unknown CuSO<sub>4</sub> solution by titrating it against intermediate sodium thiosulphate (Hypo) solution using starch as final indicator.[Iodometric Titration]
- 5. To determine the chloride content of the given sample of water using silver nitrate solution with potassium chromate solution as an indicator.[Precipitation Titration]
- 6. To determine Moisture content in given sample of coal.[Proximate Analysis]
- 7. To determine Ash content in given sample of coal. [Proximate Analysis]
- 8. To determine the Viscosity Index of give lubricating oil by Redwood Viscometer No.1 and Redwood Viscometer 2.[Lubricating Oil Analysis]
- 9. To determine the Flash Point and Fire Point of lubricating oil by Abel's Apparatus.[Lubricating Oil Analysis]
- 10. To determine the Flash Point and Fire Point of lubricating oil by Pensky Martin's Apparatus.[Lubricating Oil Analysis]
- 11. To determine S.E.N. of given lubricating oil[Lubricating Oil Analysis].
- 12. To separate mixture of pigments by Thin Layer Chromatography [Instrumental Methods].
- 13. To separate mixture of pigments by Paper Chromatography [Instrumental Methods].
- 14. To verify Beer Lambert's law of colorimetry [Instrumental Methods].
- 15. To determine amount of Iron by colorimetry [Instrumental Methods].
- 16. To estimate amount of Iron by UV spectrophotometer. [Instrumental Methods]
- 17. To determine pH of given solution using pH meter. [Instrumental Methods]
- 18. To determine strength of acid/base by conductometric titrations. [Instrumental Methods]

#### **TEXT BOOKS:**

- Engineering Chemistry Jain & Jain Dhanpat Rai & Company Pvt. Ltd, New Delhi.
- A Text Book of Engineering Chemistry S.S. Dara S. Chand Publication, Delhi.

- Engineering Chemistry- Shashi Chawla, Dhanpat Rai &Company Pvt. Ltd, Delhi.
- Engineering Chemistry Uppal Khanna Publishers.
- A Text book of Engg. Chemistry- Agarwal, C.V, Murthy C.P, Naidu, BS Publication, Hyderabad.
- B. Sivasankar, Engineering Chemistry 1 st Edition, Mc Graw Hill Education (India), 2008
- O.G. Palanna, McGraw Hill Education (India) Private Limited, 9 th Reprint, 2015

#### **REFERENCE BOOKS:**

- Chemistry in Engineering and Technology, Kuriacose J.C. and Rajaram J., Tata McGraw Hill.
- Applied Chemistry- Theory and Practice, O.P. Viramani, A.K. Narula, New Age International Pvt. Ltd. Publishers, New Delhi.
- Chemistry of Engineering Material-C.V. Agarwal, Andranaidu C. Parameswara Moorthy –B.S. Publications.
- William Kemp, Organic Spectroscopy, 3 rd edition, Palgrave, New York, 2005.

#### Modes of Evaluation and Rubric

Evaluation will be continuous as an integral part of the class as well through external assessment. Laboratory assessment will be based on assignments, presentations, and viva of each candidate.

#### List/Links of e-learning resource

- Engineering Chemistry (NPTEL Web-book), by B.L. Tembe, Kamaluddin and M.S. Krishnan
- https://nptel.ac.in/course.html
- https://iln.ieee.org/resources/e-learning
- https://www.researchgate.net/publication/221928462 ELearning Usage During Chemical Engineering Courses
- https://learncheme.com/
- https://www.anits.edu.in/elearn c.php

Recommendation by Board of studies on	14.6.2022 (Tuesday)
Approval by Academic council on	16.6.2022 (Thursday)
Subject handled by department	Applied Science (Chemistry)

Manys

Dr Manju Singh Prof & Head, Chemistry UIT, RGPV, Bhopal (Dr.NSsapre)

Dr Nitin Sapre Prof & Head, Chemistry SGSITS, Indore Jan 1

Dr J Parashar Dean, Academics SATI, Vidisha Dr Manoj Datar Prof & Head, Chemistry SATI, Vidisha



(Engineering College), VIDISHA M.P. (An Autonomous Institute Affiliated to RGPV Bhopal)

### -----CIVIL ENGINEERING-----

Semester/Y	ear		II/I		Progr	am			B. Tech			
Subject Category	ESC	Su	bject Code:	CEA101		101 Subject Name:		Basic Civil Engineering				ing
Maximum Marks Allotted										act Ho	ours	
	-	Γheor	У		F	Practical		Total	Contact Hours			Total
End Sem	Mid S	om	Assignment	Ouiz	End	Lab-	Ouiz			т	Ь	Credits
Liid Seili	IVIIU-O	CIII	Assignment	QuiZ	Sem	Work	Quiz	IVIAINS			'	
60	20		10 10 30 10 10 150		150	3	-	2	4			
End Sem	Mid-S	em	y Assignment	Quiz	End Sem	Lab- Work	Quiz	Total Marks	L	Т	Р	

#### Prerequisites:

Nil

#### Course Objective:

Students are expected to learn the basic concepts of Civil Engineering and to think clearly and critically the solution of Engineering problems with this knowledge in their respective fields.

#### Course Outcomes:

After completion of the course, the student will be able to:

- 1. Understand the basic concepts and fundamental principles of Civil Engineering.
- 2. To enhance their understanding and apply this knowledge in their specific courses for the analysis and design problems.

UNITs	Descriptions	Hrs.	CO's
I	<b>Building Materials:</b> Stones, bricks, cement, lime, timber-their types, properties, tests & uses, laboratory tests of concrete and mortar Materials: Workability and Strength properties of Concrete, Preparation of concrete, compaction, curing, etc.	8	CO1 & CO2
II	Elements of Building Construction: Various components of a building and their functions, Types of foundations, Brick masonry walls, plastering and pointing, floors, roofs, Doors, windows, lintels, staircases – types and their suitability	8	CO1 & CO2
III	Surveying & Positioning: Introduction to surveying Instruments – levels, thedolites, plane tables and related devices. Electronic surveying instruments etc. Measurement of distances – conventional and EDM methods, measurement of directions by different methods, measurement of elevations by different methods. Reciprocal levelling.	10	CO1 & CO2
IV	<b>Mapping &amp; Sensing:</b> Mapping details and contouring, Profile Cross sectioning and measurement of areas, volumes, application of measurements in quantity computations, Survey stations, Introduction of remote sensing and its applications.	7	CO1 & CO2
V	<b>Transportation Engineering:</b> Role of Transportation in National development, Transportation Ways, Surface- Transportation and Aviation, BOT & BOOT Projects for Highways, Elements of Traffic Engineering and Traffic Control	7	CO1 & CO2
-	ures (if any)		
Total Hour	"S	40	

#### Suggestive list of experiments:

- 1. To find the Consistency of cement.
- 2. To find the Initial & Final Setting time of cement.

- 3. To find the Fineness of cement.
- 4. To find the Specific Gravity of cement.
- 5. To find compressive strength of cement.
- 6. To find the Specific Gravity of sand
- 7. To find the sieve analysis and zoning of sand
- 8. To find the bulking and water absorption of sand.
- 9. Testing of coarse aggregate: Specific Gravity, sieve analysis, water absorption
- 10. To find the Water Absorption and compressive strength of Brick.

#### Text Book-

- 1. Building Construction by Sushil Kumar.
- 2. Civil Engineering materials, TTTI, Chandigarh.
- 3. Surveying Vol. I & II by Dr. B. C. Punamia Publication Laxmi Publication Delhi
- 4. Building Construction, Author: Dr. B. C. Punamia, Publisher: Laxmi Pub. Delhi
- 5. Engineering Material, Author: Dr. S.C. Rangwala, Publisher: Charotar Pub. House
- 6. Highway Engineering, Author: Khanna S. K. and Justo C. E.G., Publisher: Nemchand and Brothers

#### Reference Books-

- 1. Shesha Prakash and Mogaveer; Elements of Civil Engg & Engg. Mechanics; PHI
- 2. Building Materials by S.C. Rangwala-Charotar publications House, Anand
- 3. Building Construction by Sushil Kumar.
- 4. Surveying Vol. I and II, Author: S. K. Duggal, Publisher: Tata Macgraw hill Publication New Delhi

#### Modes of Evaluation and Rubric

Quiz, Assignment, Midterm exam, End term exam and Practical Viva.

Rubric: End term exam. Practical: 50% Quiz and 50% Viva.

#### List/Links of e-learning resource

http://www.nptel.iitm.ac.in/courses.php?branch=Civil http://www.nptel.iitm.ac.in/courses/Webcourse-contents/IIT

Recommendation by Board of studies on	13-06-2022
Approval by Academic council on	
Compiled and designed by	
Subject handled by department	Civil Engineering Department



#### (Engineering College), VIDISHA M.P.

(An Autonomous Institute Affiliated to RGPV Bhopal)

#### -----CIVIL ENGINEERING-----

Semester/Year I/I				Program			B.Tech					
Subject Category	ESC	Sub	ect Code:	CEA	103	Subje Name	II.	Engir	ineering Mechanics			
Maximum Marks Allotted									Contact Hours			
		Theory				Practical		Total	Conta	act i it	Juis	Total
End Sem	Mid-	Sem	Assign ment	Quiz	End Sem	Lab- Work	Quiz	Marks	L	Т	Р	Credits
60	20	0	10	10	30	10	10	150	3	-	2	4

#### Prerequisites:

Physics and Mathematics.

#### Course Objective:

Students are expected to learn the basic concepts of Engineering Mechanics and to think clearly and critically the solution of Engineering problems with this knowledge in their respective fields.

#### Course Outcomes:

After completion of the course, the student will be able to:

- 1. Understand the basic concepts and fundamental principles of Engineering Mechanics.
- 2. To enhance their understanding and apply this knowledge in their specific courses for the analysis and design problems.

UNITs	Descriptions	Hrs.	CO's			
I	<b>Equilibrium of System of Forces:</b> Force Systems Basic concepts, Particle and Rigid Body equilibrium; Coplanar-Concurrent and Nonconcurrent Forces, Components in Space, Resultant, Moment of Forces and its Applications; Equilibrium of System of Forces, Free body diagrams, Equations of Equilibrium of Coplanar Systems; Static Indeterminacy, Friction-Application problems such as Impending motion of connected bodies, ladder friction & belt drives.	9	CO1 & CO2			
II	<b>Trusses:</b> Introduction to various types of Trusses, Analysis of forces in the members of a Perfect truss: Method of joints, Method of Section, Graphical Methods.	7	CO1 & CO2			
III	Analysis of Beams and Simple Frames: Types of Beams, loading and supports; Shear Force, Bending moment, Axial Force diagrams for various types of determinate beams and frames.					
IV	Centroid and Moment of Inertia: Centroid of simple figures from first principles, centroid of composite sections; Moment of inertia of plane sections from first principles, Moment of inertia of standard sections and composite sections, Product of Inertia, Principal Moment of Inertia.	9	CO1 & CO2			
V	8	CO1 & CO2				
	ures (if any)					
Total Hour	'S	40				

#### Suggestive list of experiments:

- 1. To verify law of Polygon of forces.
- 2. To find the reaction at the supports of a Simply Supported Beam and verify the law of Superposition of Forces.
- 3. To determine the Coefficient of friction between different surfaces using a horizontal plane.
- 4. To find the Coefficient of friction between Rope and Drum.

- 5. To verify Shear Force at a given section of a Simply Supported Beam.
- 6. To verify Bending Moment at a given section of a Simply Supported Beam.

#### Text Book-

- 1. Shesha Prakash and Mogaveer; Elements of Civil Engg & Engg. Mechanics; PHI
- 2. Civil Engineering materials, TTTI, Chandigarh.
- 3. R.C. Hibbler- Engineering Mechanics: Statics & Dynamics

#### Reference Books-

- 1. Engineering Mechanics by R.K. Bansal
- 2. Beer & Johnston, Vector Mechanics for Engineering-Statics & Dynamics.
- 3. Engineering Mechanics by Bhavi Katti

#### Modes of Evaluation and Rubric

Quiz, Assignment, Midterm exam, End term exam and Practical Viva.

Rubric: End term exam. Practical: 50% Quiz and 50% Viva.

#### List/Links of e-learning resource

https://nptel.ac.in/courses/122/104/122104015/

https://nptel.ac.in/courses/105/106/105106116/

https://nptel.ac.in/courses/105/106/105106201/

Recommendation by Board of studies on	13-06-2022
Approval by Academic council on	
Compiled and designed by	
Subject handled by department	Civil Engineering Department



(Engineering College), VIDISHA M.P.

(An Autonomous Institute Affiliated to RGPV Bhopal)

#### **Department of Humanities and Management**

	Semester/Ye	ear	1/11		Program		B. Fech.					
	Subject Category	Hum	Subject Code:	HUB10	12	oject me:	Communica	tion ar	riting			
		Maximum Marks Allotted										
		Theo	ory		Prac	ctical -	Total	Con	tact H	Juis	Total	
	End Sem	Mid-Sem	Quiz	Assign ment	End Sem	Lab-Work	Marks	L	Т	Р	Credits	
İ	60	20	10	10	-	-	100	3	-	2	4	

#### Prerequisites:

In this era of Globalization and Information Technology, English has a special and predominant role in the communicative sphere and thus English commands the most prestigious position in the world in the exchange of information across geographical boundaries. The syllabus has been designed to develop linguistic and communicative competence of Engineering Students.

#### Course Objective:

- To improve the language proficiency of the students in English with emphasis on LSRW Skills.
- 2. To enable the students to study and comprehend the prescribed lessons and subjects more effectively relating to their theoretical and practical components.
- 3. To develop the communication skills of the students in both formal and informal situations.

#### Course Outcomes:

- 1. Students will develop the ability to listen, speak, read and write effectively in both academic and non-academic environment.
- 2. The students will have an understanding of multidisciplinary contexts.
- 3. They will be able to successfully handle real life situations of business correspondence.
- 4. They will also develop the ability to analyse and interpret any technology related subjects.
- 5. Students will be in a position to make presentations on topics of technical and general interests; current issues related to politics; work and business environment.

UNITs	Descriptions	Hrs.	CO's	
I	Significance of Communication: Process of Communication, The importance of Effective Communication in Business, Verbal and Non-Verbal Communication, Oral and Written Communication, Barriers to Communication.	10	1	
l II	Employability Traits: Job Interview (Body Language), Types of Interviews, Interview Skills, Employability Skills, Group Discussion.	6	2	
III	III Soft Skills: Goal Setting, Qualities of a good leader, Time Management, Time Wasters, Problem Solving.			
IV	8	4		
V	8	5		
Guest Lect				
Total Hour	rs	40		

#### Suggestive list of experiments: NA

#### 1. NA

#### Text Book-

1. A.J. Thomson and A.V. Martinet, A Practical English Grammar, Oxford IBH Pub Sanjay Kumar PushpLata, English for Effective Communication, Oxford.

#### Reference Books

- Language and Life: A Skills Approach Board of Editors, Orient Black Swan Publishers,
- India. 2018.
- 3. Business Correspondence and Report Writing By R C Sharma; TMH.
- 4. Living English Structure By W.S. Allen; Longmans.
- 5. English Grammar Ehrlich, Schaum Series; TMH.

- 6. Spoken English for India By R.K. Bansal and IB Harrison Orient Longman.
- 7. New International Business English by Joans and Alexander; OUP.
- 8. Effective Technical Communication Rizvi; TMH
- 9. Body Language Vinay Mohan Sharma

#### Modes of Evaluation and Rubric

Two mid semester tests, Quiz, Sessional an end semester examination.

#### List/Links of e-learning resource

- https://onlinecourses.nptel.ac.in
- https://www.classcentral.com (swayam)

Recommendation by Board of studies on	26/02/2022
Approval by Academic council on	
Compiled and designed by	Dr. Amitish Singh, Dr. Manorama Saini and Dr. Veena Datar
Subject handled by department	Department of Humanities











# SAMRAT ASHOK TECHNOLOGICAL INSTITUTE (Engineering College), VIDISHA M.P.

(An Autonomous Institute Affiliated to RGPV Bhopal)

#### **Department of Applied Science**

Semester/Year First Sem Program				B.Te	ch.					
Subject	Departmen	t Subje	Subject		Subject	Lincor A	Jachro	and C	`oloulı	10
Category	al Core	Code	e:   IVI	AB101	Name:	Linear A	ligebia	anu C	aicuit	15
Maximum Marks A				Allotted			Cont	act Ho	oure	
	Theory			Theory Practical			Con	actin	Juis	Total
End Sem	Mid-Sem Quiz	Assign	End	Lab-	Total Marks		т	Р	Credits	
Liid Seili	Wild-Selfi	QuiZ	ment	Sem	Work		_	'		
60	20	10	10	-	-	100	3	1	-	4

#### Prerequisites:

Basic of Differentiations, Integrations and Matrices.

#### Course Objective:

The objective of this course is to familiarize the prospective engineers with techniques in calculus, and linear algebra. It aims to equip the students with standard concepts and tools at an intermediate to advanced level that will serve them well towards tackling more advanced level of mathematics and applications that they would find useful in their disciplines.

#### Course Outcomes:

This course is to develop students abilities to:

- 1. Apply Differential Calculus to Notions of Curvature. Apart from some other Applications they will have a Basic Understanding of Taylor's Theorem, Maxima and Minima.
- 2. The Fallouts of Partial Differentiation that is Fundamental to Application of Analysis to Engineering Problems.
- 3. Finding area and Volume using Double and Triple Integrals.
- 4. The Essential Tool of Matrices and Linear Algebra in a Comprehensive Manner. Student will understand Matrices and their Application to Solve System of Linear Simultaneous Equations.
- 5. Students will Gain Experience with Problem Solving in Boolean Algebra and Graph Theory.

UNITs	Descriptions	Hrs.	CO's
I	<b>Differential Calculus:</b> Lebnitz Theorem, Expansion of functions by Maclaurins and Taylors theorem (one variable), Maxima & Minima of two variables, Curvature: Radius and Centre of Curvature for Cartesian Coordinates.	8	1
11	Partial Differentiation: Partial Derivatives of Higher Order, Homogeneous Functions, Euler's Theorem, Total differentiation, Errors and Approximations.	8	2
III	Integral Calculus: Definite Integral as a Limit of the Sum, Application in Summation of Series, Multiple Integrals, Change of order of Integration, Application of Double and Triple Integrals (Area & Volume).	8	3
IV	<b>Matrix</b> : Definition, Types & Properties of Matrices, Elementary Transformation, Rank of Matrix, Consistency of Linear System of Equations and their solutions, Eigen Values and Eigen Vectors, Cayley Hamilton Theorem and its Application to find the Inverse.	8	4

	Boolean Algebra & Graph Theory: Algebra of logic, Principal of		
V	Duality and basic theorem, Boolean expression and Boolean functions, Definition of Graph, Types of Graphs, Sub Graphs, Walk, Path and	8	5
	Circuits,.		
TOTAL HOURS			

#### Reference Books:

Engg. Mathematics: By B.S. Grewal
 Boolean Algebra: R.S. Agrawal
 Engg. Mathematics: by H.K. Dass

4. Engg. Mathematics : By B. V. Rammanna

Recommendation by Board of studies on	14-06-2022
Approval by Academic council on	16-06-2022
Compiled and designed by	Applied Maths Board of Studies, Chairman Dr. Shailesh Jaloree

A

Juil leup



(Engineering College), VIDISHA M.P.

(An Autonomous Institute Affiliated to RGPV Bhopal)

#### **Department of Humanities and Management**

Semester/Ye	ear	II Year	Program				B.	3. Tech All Branches			
Subject	MAC	Subject	NAAC	C101	Subj	ect	Univ	ersal H	uman	Value	es
Category	IVIAC	Code:	IVIAC	5101	Nam	ne:					
	Maximum Marks Allotted						Cont	act H	oure		
	Theory			Practical			Total	Com	actin	Juis	Total
End Sem	m Mid S	Mid-Sem Quiz	Assign	End	Lab-	Quiz	Marks	1	т	Р	Credits
Liid Seiii	IVIIU-SE	JIII QUIZ	ment	Sem	Work	Quiz	IVIAINS	_	' '	-	
00	00	00	00	60	20	20	100	-	-	2	Grade

#### Prerequisites:

During the Induction Program, students would get an initial exposure to human values through Universal Human Values – I. This exposure is to be augmented by this compulsory full semester foundation course.

#### Course Objective:

At the end of the course, the students will be able to:

- 1. Develop a holistic perspective based on exploration about others and themselves.
- 2. Develop clarity, importance of harmony and humanity towards family, society and nature/existence.
- 3. Strengthen self-reflection.
- 4. Develop commitment and courage to act.

#### Course Outcomes:

- 1. By the end of the course, students will become aware of themselves, and their surroundings (family, society, nature)
- 2. They would have better critical ability.
- 3. They would become more responsible in life; and keeping human relationships and human nature in mind will be able to handle problems with sustainable solutions.
- 4. They would also become sensitive to their commitment towards nature and existence.
- 5. They would be able to apply what they have learnt to their own selves in different day-to-day reallife scenarios, at least a beginning would be made in this direction.

UNITs	Descriptions	Hrs.	CO's
I	Introduction - Need, Basic Guidelines, Content and Process for Value Education  1. Self-Exploration—what is it? - Its content and process; 'Natural Acceptance' and Experiential Validation- as the process for self-exploration  2. Continuous Happiness and Prosperity- A look at basic Human Aspirations  3. Right understanding, Relationship and Physical Facility- the basic requirements for fulfilment of aspirations of every human being with their correct priority  4. Understanding Happiness and Prosperity correctly- A critical appraisal of the current scenario	8	1

	5. Method to fulfil the above human aspirations: understanding and		
	living in harmony at various levels. Include practice sessions to discuss		
	natural acceptance in human being as the innate acceptance for living		
	with responsibility.		
	Understanding Harmony in the Human Being - Harmony in Myself!		
	1. Understanding human being as a co-existence of the sentient 'l' and		
	the material 'Body'		
	2. Understanding the needs of Self ('I') and 'Body' - happiness and		
	physical facility		
	3. Understanding the characteristics and activities of 'I' and harmony in	_	
II	in the second se	6	2
	4. Understanding the harmony of I with the Body: Sanyam and Health;		
	correct appraisal of Physical needs, meaning of Prosperity in detail		
	5. To ensure Sanyam and Health. Include practice sessions to discuss		
	the role others have played in making material goods. Identifying from		
	one's own life. Differentiate between prosperity and accumulation.		
	Understanding Harmony in the Family and Society- Harmony in		
	Human- Human Relationship		
	1. Understanding values in human-human relationship; meaning of		
	Justice (nine universal values in relationships) and program for its		
	fulfillment to ensure mutual happiness.		
	2. Understanding the meaning of Trust; Difference between intention		
	and competence.		
III	3.Understanding the meaning of Respect, Difference between Respect	4	3
	and differentiation; the other salient values in relationship.		
	4.Understanding the harmony in the society (society being an extension		
	of family): Resolution, Prosperity, fearlessness (trust) and co-existence		
	as comprehensive Human Goals.		
	5. Visualizing a universal harmonious order in society- Undivided		
	Society, Universal Order- from family to world family. Gratitude as a		
	universal value in relationships. Elicit examples from students' lives.		
	Understanding Harmony in the Nature and Existence - Whole existence		
	as Coexistence		
	1. Understanding the harmony in the Nature.		
	2. Interconnectedness and mutual fulfilment among the four orders of		
	nature recyclability and self-regulation in nature.		
IV	3. Understanding Existence as Co-existence of mutually interacting	8	4
	units in all-pervasive space.		•
	4. Holistic perception of harmony at all levels of existence.		
	5. Include practice sessions to discuss human being as cause of		
	imbalance in nature (film "Home" can be used), pollution, depletion of		
	resources and role of technology etc.		
	Implications of the above Holistic Understanding of Harmony on		
	Professional Ethics		
	Natural acceptance of human values.		
	Natural acceptance of numeri values.     Definitiveness of Ethical Human Conduct.		
V	3. Basis for Humanistic Education, Humanistic Constitution and	9	5
	Humanistic Universal Order		
	4. Competence in professional ethics: a. Ability to utilize the		
	professional competence for augmenting universal human order b.		
	Ability to identify the scope and characteristics of people friendly and		
	eco-friendly production systems, c. Ability to identify and develop		

appropriate technologies and management patterns for above		
production systems.		
5. Strategy for transition from the present state to Universal Human		
Order: a. as socially and ecologically responsible engineers,		
technologists b. At the level of society: as mutually enriching institutions		
and organizations.		
Guest Lectures (if any)	5	
Total Hours	40	

#### Suggestive list of experiments:

Text Book-Human Values and Professional Ethics by R R Gaur, R Sangal, G P Bagaria, Excel Books, New Delhi, 2010

#### Reference Books-

- 1. JeevanVidya: EkParichaya, A Nagaraj, JeevanVidyaPrakashan, Amarkantak, 1999.
- 2. Human Values, A.N. Tripathi, New Age Intl. Publishers, New Delhi, 2004.

#### Modes of Evaluation and Rubric

Questionnaire, Quiz, Presentation and standard procedure will be followed.

#### List/Links of e-learning resource

https://fdp-aicte-india.org

https://vvce.ac.in

Recommendation by Board of studies on	26/02/2022
Approval by Academic council on	
Compiled and designed by	Dr. Manorama Saini and Dr. VeenaDatar
Subject handled by department	Humanities and Management







